

**pH-Sensitive Mucoadhesive Film-Forming Gels and Wax-Film Composites  
Suitable for Topical and Mucosal Delivery of Molecules**

**FIELD OF THE INVENTION**

The present invention relates to compositions and methods to treat the skin and mucosal surfaces with mucoadhesive film-forming gels and wax-film composites that are pH-sensitive. The invention also relates to using such pH-sensitive film-forming gels and wax-film composites to deliver molecules of interest, namely active pharmaceuticals. The invention also relates to the use of said pH-sensitive film-forming gels and wax-film composites to deliver molecules of interest, such as small molecules, peptides, proteins, and nucleic acids either locally to act at the site of administration or for the absorption of said molecules of interest across biological membranes into the systemic circulation.

**BACKGROUND OF THE INVENTION**

Pharmaceutical formulations intended to deliver drugs topically, either for local action at the site of administration or for absorption into the systemic circulation, have been described in the literature. These formulations are designed either for adherence of the dosage form to the skin (for dermal or transdermal delivery of drugs) or mucosal surface (for mucosal or transmucosal delivery of drugs). For skin delivery, formulations generally consist of dermal patches, pastes, band-aids, gels, lotions, sprays, or creams. For mucosal delivery, formulations generally consist of gels, creams, tablets, sprays, or films. In either case, one specific requirement is that the dosage form remains at the administration site for a sufficient amount of time so that the drug may function as needed. A second specific requirement is that the dosage form must consist of pharmaceutically acceptable materials. Many of the reported and commercially available delivery systems intended for topical and mucosal delivery are aqueous-based formulations comprising water-soluble excipients. However, these systems tend to be easily and quickly washed away from the application

site within minutes after application. This is very undesirable if the drug must remain at the application site for a prolonged period in order to be efficacious. In an attempt to prolong the residence time at the application site, researchers have described the use of water-insoluble excipients in the formulations. However, most often the use of water-insoluble excipients necessitates the use of non-aqueous solvents to dissolve the excipients. Excessive and repeated administration of non-aqueous solvents is not desirable.

Different types of formulations to deliver drugs topically are known in the art. Specific examples are illustrated below.

U.S. Patent 4,715,369 by Suzuki *et al.* relates to methods to treat the injured oral mucosa with a thin two-layer tablet comprised of both an adhesive layer and a nonadhesive layer. The adhesive layer is comprised of water-soluble cellulose-based polymers and polyacrylic acid while the nonadhesive layer is comprised mostly of materials with no adhesive properties. The Suzuki patent does not teach the use of film-forming gels comprised of pH-sensitive polymers and water-insoluble mucoadhesive polymers that form films due to changes in pH and/or desolvation of the polymers. Further, the Suzuki patent does not teach the use of pH-sensitive wax-film composites intended to remain adhered to the skin or mucosal surface for a prolonged period of time.

U.S. Patent 5,800,832 by Tapolsky *et al.* relates to a water-soluble, bioerodable device for the delivery of drugs, and specifically dyclonine, to mucosal surfaces. The device comprises an adhesive layer as well as a non-adhesive layer. Both the adhesive layer and the nonadhesive layer consist of water-soluble polymers such as cellulose-based polymers. The Tapolsky patent does not teach the use of film-forming gels comprised of pH-sensitive polymers and water-insoluble mucoadhesive polymers that form films due to changes in pH and/or desolvation of the polymers. Further, the Tapolsky patent does not teach the use of pH-sensitive wax-film composites intended to remain adhered to the skin or mucosal surface for a prolonged period of time.

U.S. Patent 5,955,097 by Tapolsky *et al.* relates to a non water-soluble gel, which adheres to mucosal surfaces and body tissues upon application and forms a film. The gel comprises at least one water-insoluble cellulose-based polymer, a non-aqueous solvent, and at least one active pharmaceutical. The patent teaches that upon application to the mucosal

surface or skin, the non-aqueous solvent, primarily ethanol, evaporates, diffuses, or penetrates the surrounding tissue, resulting in precipitation of the non water-soluble polymers into films. The Tapolsky patent further teaches the use of 50 to 80% ethanol by weight in the pharmaceutical gel. The Tapolsky patent does not teach the use of film-forming gels containing at least 25% water by weight and comprised of pH-sensitive polymers and water-insoluble mucoadhesive polymers that form films due to changes in pH and/or desolvation of the polymers. Further, the Tapolsky patent does not teach the use of pH-sensitive wax-film composites intended to remain adhered to the skin or mucosal surface for a prolonged period of time.

U.S. Patent 5,780,045 by McQuinn *et al.* relates to a transmucosal drug delivery device in the form of a sheet comprising an acid-containing particulate polymer dispersed in a polytetrafluoroethylene support matrix. The McQuinn patent does not teach the use of film-forming gels comprised of pH-sensitive polymers and water-insoluble mucoadhesive polymers that form films due to changes in pH and/or desolvation of the polymers. Further, the McQuinn patent does not teach the use of pH-sensitive wax-film composites intended to remain adhered to the skin or mucosal surface for a prolonged period of time.

U.S. Patent 4,552,751 by Inaba *et al.* relates to the preparation of multi-layered films of three, five, or seven layers for the administration of specific prostaglandins to mucosal sites. The Inaba patent does not teach the use of film-forming gels comprised of pH-sensitive polymers and water-insoluble mucoadhesive polymers that form films due to changes in pH and/or desolvation of the polymers. Further, the Inaba patent does not teach the use of pH-sensitive wax-film composites intended to remain adhered to the skin or mucosal surface for a prolonged period of time.

U.S. Patent 4,517,173 by Kizawa *et al.* relates to a film preparation consisting of at least three layers, including a pharmaceutical layer, a poorly water-soluble layer, and an intermediate layer. The pharmaceutical layer consists of prednisolone and allantoin together with a water-soluble cellulose-based polymer. The poorly water-soluble layer consists of shellack or fatty acids. The Kizawa patent does not teach the use of film-forming gels comprised of pH-sensitive polymers and water-insoluble mucoadhesive polymers that form films due to changes in pH and/or desolvation of the polymers.

Further, the Kizawa patent does not teach the use of pH-sensitive wax-film composites intended to remain adhered to the skin or mucosal surface for a prolonged period of time.

U.S. Patent 5,192,802 by Rencher describes the use of a bioadhesive teething gel comprising benzocaine, sodium carboxy methyl cellulose; an agent selected from the group consisting of xanthan gum and sodium alginate and a diluent selected from the group consisting of polyethylene glycol and polyethylene glycol with glycerine. U.S. Patent 5,314,915 by Rencher and U.S. Patent 5,298,258 by Akemi *et al.* also describe the use of aqueous or oil-based bioadhesive gelling agents. These patents do not teach the use of film-forming gels comprised of pH-sensitive polymers and water-insoluble mucoadhesive polymers that form films due to changes in pH and/or desolvation of the polymers. Further, these patents do not teach the use of pH-sensitive wax-film composites intended to remain adhered to the skin or mucosal surface for a prolonged period of time.

U.S. Patents 5,081,157 and 5,081,158 by Pomerantz relate to a film-forming composition for topical application of medicaments to body tissues. The film-forming composition includes hydroxypropyl cellulose, a volatile solvent, and an esterification agent which reacts with the hydroxypropyl cellulose to form a reaction product which is soluble in the solvent. These patents do not teach the use of film-forming gels comprised of pH-sensitive polymers and water-insoluble mucoadhesive polymers that form films due to changes in pH and/or desolvation of the polymers. Further, these patents do not teach the use of pH-sensitive wax-film composites intended to remain adhered to the skin or mucosal surface for a prolonged period of time.

U.S. Patent 4,900,554 by Yanagibashi *et al.* relate to the use of a device for the delivery of drugs in the oral cavity. The device comprises an adhesive layer consisting of at least one acrylic acid polymer, a water-insoluble cellulose derivative, and a pharmaceutical preparation, and a water-insoluble or sparingly soluble backing layer. Yanagibashi *et al.* state that "it is impossible to achieve an adhesive device for application to body tissue without all three components, that is, acrylic acid polymer, water insoluble cellulose derivative and water insoluble or sparingly soluble backing layer". The Yanagibashi patent does not teach the use of film-forming gels comprised of pH-sensitive polymers and water-insoluble mucoadhesive polymers that form films due to changes in

pH, and/or desolvation of the polymers. Further, the Yanagibashi patent does not teach the use of pH-sensitive wax-film composites intended to remain adhered to the skin or mucosal surface for a prolonged period of time.

As illustrated, the references described above appear to lack preferred compositions and properties for an efficacious and pharmaceutically acceptable bioadhesive delivery system. Namely, the references do not describe primarily aqueous-based film-forming gels comprised of pH-sensitive polymers and water-insoluble mucoadhesive polymers that form films due to changes in pH and/or desolvation of the polymers. Further, the references do not describe pH-sensitive mucoadhesive wax-film composites intended to remain adhered to the skin or mucosal surface for a prolonged period of time.

### **SUMMARY OF THE INVENTION**

The invention is directed to a pharmaceutical gel composition comprising:

- a solvent vehicle,
- at least one water-insoluble swellable mucoadhesive polymer,
- at least one pH-sensitive film-forming polymer, and
- at least one molecule of interest.

The solvent vehicle may be comprised of at least 25 to 100 parts water or buffered water with 0 to 75 parts of ethanol, propylene glycol, glycerin, polyethylene glycol, or combinations thereof. The water-insoluble swellable mucoadhesive polymer may be polyacrylic acid cross-linked with polyalkenyl ether or divinyl glycol, wherein the water-insoluble swellable mucoadhesive polymer is preferably, Noveon or Carbomer. The water-insoluble swellable mucoadhesive polymer may be present at a concentration of from 0.1% to 20% by weight.

The pH-sensitive polymer is a copolymer of methacrylic acid and acrylic or methacrylic ester. Preferably, it is present at a concentration of from 0.05% to 10% by weight. More preferably, the polymer is a Eudragit polymer, or a chemical derivative thereof.

The molecule of interest may comprise an active pharmaceutical such as an antimicrobial, antiviral, antiinflammatory, antiseptic, antihistamine, a local anesthetic, a disinfectant, a keratolytic, an analgesic, an anti-migraine, an anti-fungal, a sweetener, a flavoring agent, a diagnostic agent, or combination thereof. Preferably, the molecule of interest is amlexanox, triclosan, peptide, protein, hirudin, plasmid DNA, lidocaine, benzocaine, dyclonine, or benzodiazepine drug or a derivative thereof.

The invention is also directed to a pharmaceutical gel which when applied to the skin or mucosal surface forms a film, said gel comprising a solvent vehicle, at least one water-insoluble swellable mucoadhesive polymer, at least one pH-sensitive film-forming polymer, and at least one molecule of interest, wherein said film is formed due to changes in pH and desolvation of the polymer, and wherein said film provides for the delivery of the molecule of interest to or through the application site. The solvent vehicle may be comprised of at least 25 to 100 parts water with 0 to 75 parts of ethanol, propylene glycol, glycerin, polyethylene glycol, or combinations thereof. The water-insoluble swellable mucoadhesive polymer may be polyacrylic acid cross-linked with polyalkenyl ether or divinyl glycol. Preferably, the water-insoluble swellable mucoadhesive polymer is Noveon or Carbomer. The water-insoluble swellable mucoadhesive polymer may be present at a concentration of from 0.1% to 20% by weight.

The pH-sensitive polymer may be a copolymer of methacrylic acid and acrylic or methacrylic ester. Preferably, it is present at a concentration of from 0.05% to 10% by weight. More preferably, the polymer is a Eudragit polymer, or a chemical derivative thereof.

The molecule of interest may comprise an active pharmaceutical such as an antimicrobial, antiviral, antiinflammatory, antiseptic, antihistamine, a local anesthetic, a disinfectant, a keratolytic, an analgesic, an anti-migraine, an anti-fungal, a sweetener, a flavoring agent, a diagnostic agent, or combination thereof. Preferably, the molecule of interest is amlexanox, triclosan, peptide, protein, hirudin, plasmid DNA, lidocaine, benzocaine, dyclonine, or benzodiazepine drug or a derivative thereof. The application site may be the skin, mouth, vagina, nose, nasal cavity, or other accessible mucosal site.

The invention is also directed to a wax-film composite comprised of a pH-sensitive mucoadhesive layer and a water-insoluble wax layer. The pH-sensitive mucoadhesive layer may be present at a concentration of 20% to 90% by weight, and the water-insoluble wax layer may be present at a concentration of 10% to 80% by weight. The pH-sensitive mucoadhesive water-insoluble layer may be comprised of:

- at least one water-insoluble swellable mucoadhesive polymer,
- at least one pH-sensitive film-forming polymer, and
- at least one molecule of interest.

The water-insoluble swellable mucoadhesive polymer is polyacrylic acid cross-linked with polyalkenyl ether or divinyl glycol. Preferably, the water-insoluble swellable mucoadhesive polymer is Noveon or Carbomer. The water-insoluble swellable mucoadhesive polymer may be present in the pH-sensitive mucoadhesive layer at a concentration of from 0.1% to 20% by weight.

The pH-sensitive polymer may be a copolymer of methacrylic acid and acrylic or methacrylic ester. Preferably, it is present at a concentration of from 0.05% to 10% by weight. More preferably, the polymer is a Eudragit polymer, or a chemical derivative thereof.

The molecule of interest may comprise an active pharmaceutical such as an antimicrobial, antiviral, antiinflammatory, antiseptic, antihistamine, a local anesthetic, a disinfectant, a keratolytic, an analgesic, an anti-migraine, an anti-fungal, a sweetener, a flavoring agent, a diagnostic agent, or combination thereof. Preferably, the molecule of interest is amlexanox, triclosan, hirudin, plasmid DNA, lidocaine, benzocaine, dyclonine, or benzodiazepine drug or a derivative thereof. The wax-film composite may comprise at least one water-insoluble pharmaceutical wax having a melting point between 40°C and 100°C and at least one water-soluble or water-swellable polymer.

The wax may be DENTSPLY® Utility Wax, beeswax, emulsifying wax, microcrystalline wax, carnauba wax, paraffin wax, white wax, yellow wax, or other suitable pharmaceutical wax. The water-soluble or swellable polymer may be present in the insoluble wax layer at a concentration of 0.05 to 10% by weight. The water-soluble or swellable polymer may be tragacanth, polyvinyl pyrrolidone, polyvinyl alcohol, cross-

linked polyacrylic acid, polyethylene glycol, a cellulose polymer derivative, or other suitable pharmaceutical polymer that is water-soluble or water-swellaable. In the wax-film composite, the molecule of interest may be contained in and released from either the pH-sensitive mucoadhesive layer or the water-insoluble wax layer.

The molecule of interest comprises an active pharmaceutical compound such as an antimicrobial, antiviral, antiinflammatory, antiseptic, antihistamine, a local anesthetic, a disinfectant, a keratolytic, an analgesic, an anti-migraine, an anti-fungal, a sweetener, a flavoring agent, a diagnostic agent, or a combination thereof. The molecule of interest may comprise an active pharmaceutical such as an antimicrobial, antiviral, antiinflammatory, antiseptic, antihistamine, a local anesthetic, a disinfectant, a keratolytic, an analgesic, an anti-migraine, an anti-fungal, a sweetener, a flavoring agent, a diagnostic agent, or combination thereof. Preferably, the molecule of interest is amlexanox, triclosan, protein, peptide, hirudin, plasmid DNA, lidocaine, benzocaine, dyclonine, or benzodiazepine drug or a derivative thereof. Preferably, the hirudin may be complexed with a substance of opposite charge. More preferably, the substance of opposite charge may be chitosan or protamine. Alternatively, the molecule of interest may be a plasmid DNA or plasmid DNA complexed with a substance of opposite charge such as chitosan, protamine, or a cationic lipid.

The application site for the wax-film composite may be the skin, mouth, vagina, nasal cavity, or other accessible mucosal site. Preferably, the wax-film composite adheres to the application site for at least 1 hour. Preferably, the wax-film composite has a total thickness of less than 5 mm.

The invention is also directed to a method of making the above pharmaceutical gel composition, comprising:

- (i) adding a mucoadhesive polymer to a stirring water to form a solution that is clear and viscous,
- (ii) adding the pH-sensitive film-forming polymer to the solution formed in step (i) and measuring the pH of the solution,
- (iii) adding the molecule of interest to the solution of step (ii), and
- (iv) forming the final gel composition.



The invention is also directed to a method of making the above wax-film composite, comprising:

- (i) forming a homogeneous mucoadhesive film by fusing a mucoadhesive polymer and a pH-sensitive film-forming polymer,
- (ii) homogeneously coating one side of the mucoadhesive film of step (i) with a melted wax composition, and
- (iii) drying the wax composition to form the wax-film composite, wherein the molecule of interest is added in either step (i) or (ii), or the molecule of interest is added to the mucoadhesive side after the wax-film composite is formed.

Furthermore, the invention is also directed to a method of treating a disease or an illness by administering to a person in need thereof, the gel or the wax-film composite as described above comprising a molecule of interest.

### BRIEF DESCRIPTION OF THE DRAWINGS

The present invention will become more fully understood from the detailed description given hereinbelow, and the accompanying drawings which are given by way of illustration only, and thus are not limitative of the present invention, and wherein:

Figure 1 shows a summary of the compositions of the invention

Figure 2 shows an *in vitro* adhesion time of ¼ inch wax-film composites (n=3 each) on glass submerged in 40 mM KH<sub>2</sub>PO<sub>4</sub>/NaOH buffer, pH 6 at 37°C (unstirred). Each mucoadhesive layer contains 5.0-5.3 mg of total polymer comprised of Noveon and Eudragit S100 in the ratios indicated. The wax layer consists of DENTSPLY® Utility Wax containing 1% w/w tragacanth polymer. \*Indicates that different volumes of mucoadhesive gels were cast as described in Example 8. See Table 1 and Example 8 for additional details.

Figure 3 shows *in vitro* adhesion time of ¼ inch wax-film composites (n=3 each) on glass submerged in 40 mM KH<sub>2</sub>PO<sub>4</sub>/NaOH buffer, pH 6 at 37°C (stirred at 100 rpms). Each mucoadhesive layer contains from 1.2 mg to 11.2 mg of total polymer comprised of Noveon and Eudragit S100 in a weight ratio of 3:1. The wax layer consists of

DENTSPLY® Utility Wax containing 1% w/w tragacanth polymer. See Table 2 and Example 9 for additional details.

Figure 4 shows *in vitro* adhesion time of ¼ inch wax-film composites containing plasmid DNA (10 µg) either incorporated during the manufacturing process ('preload') or aliquoted onto the mucoadhesive layer after the manufacturing process ('postload'). Wax-film composites (n=5 each) were adhered to glass and submerged in 40 mM KH<sub>2</sub>PO<sub>4</sub>/NaOH buffer, pH 6 at 37°C (unstirred). See Example 10 for additional details.

Figure 5 shows a standard curve for plasmid DNA in solution as measured by the PicoGreen DNA Quantitation Kit (from Molecular Probes, Inc., Eugene, OR).

Figure 6 shows the release of plasmid DNA pre-loaded into wax-film composites. Wax-film composites were made as described in Example 10 using a mucoadhesive gel comprised of Noveon/Eudragit S100 (3:1 w/w) and plasmid DNA. The wax layer consists of DENTSPLY® Utility Wax containing 1% w/w tragacanth polymer. Five ¼ inch wax-film composites containing of plasmid DNA (5 µg) were submerged separately into 1 mL 10 mM PBS buffer, pH 7.4 at 37°C. At various times, exactly 100 µL solution was aliquoted for DNA quantitation using the PicoGreen DNA Quantitation Kit. Exactly 100 µL fresh PBS was added to replace the removed volume at each time point.

Figure 7 shows the release of plasmid DNA post-loaded into wax-film composites. Wax-film composites were made as described in Example 10 using a mucoadhesive gel comprised of Noveon/Eudragit S100 (3:1 w/w). The wax layer consists of DENTSPLY® Utility Wax containing 1% w/w tragacanth polymer. Plasmid DNA (5 µg) was added to five individual ¼ inch wax-film composites, allowed to air dry for 4 hours, and then submerged separately into 1 mL 10 mM PBS buffer, pH 7.4 at 37°C. At various times, exactly 100 µL solution was aliquoted for DNA quantitation using the PicoGreen DNA Quantitation Kit. Exactly 100 µL fresh PBS was added to replace the removed volume at each time point.

Figure 8 shows: (A) The reaction kinetics of Chromozym-TH and uncomplexed thrombin. Hirudin was added to excess thrombin and residual uncomplexed thrombin was assayed spectroscopically after the addition of Chromozym-TH. Hirudin was added to the 1.0 ml volume in the following amounts: 0 ng (□), 16.9 ng (●), 33.8 ng (Δ), 50.7 ng (▲),

67.6 ng (O), and 84.4 ng (■). (B) Calibration curve for released hirudin in solution. The figure was made by plotting the difference between the thrombin base value absorbance (thrombin in the presence of substrate only) and the hirudin sample values versus the concentration of hirudin standards from A.

Figure 9 shows an HPLC Standard Curve for Hirudin. Three standard curves for hirudin were generated using a C18 Nucleosil® 5 µm analytical column (4.6 x 250 mm). The mobile phase consisted of water:acetonitrile:trifluoroacetic acid (59.95:40:0.05 w/w/w). The flow rate was 0.6 mL/min. Each 50 µL sample of each standard curve was injected three times and the average AUC was plotted.

Figure 10 shows the Particle Size of Hirudin/Chitosan Complexes in water made with hirudin at a Concentration of 0.5 mg/mL.

Figure 11 shows the Zeta Potential of Hirudin/Chitosan Complexes in water made with Hirudin at a Concentration of 0.1 mg/mL. Zeta Potential values are reported as the mean and standard deviation of three measurements.

Figure 12 shows the stability of Hirudin/Chitosan Complexes (4:1 w/w) made in water as function of Hirudin concentration and storage time. All samples were stored at 4°C. For the Day 7 sample analysis, exactly 100 µL of Hirudin/Chitosan complexes in water was diluted in 900 µL ethanol to determine the stability of the complexes in ethanol.

## **DETAILED DESCRIPTION OF THE INVENTION**

The invention relates to delivery systems suitable for the topical and mucosal delivery of molecules of interest. Specifically, the invention relates to; 1) compositions of pH-sensitive mucoadhesive film-forming gels, 2) uses of pH-sensitive mucoadhesive film-forming gels, 3) compositions of pH-sensitive mucoadhesive wax-film composites, and 4) uses of pH-sensitive mucoadhesive wax-film composites. A summary of the compositions of the invention is shown in Figure 1.

The pH-sensitive mucoadhesive film-forming gel is a pharmaceutical gel which when applied to the skin or mucosal surface forms a film, said gel comprising a solvent vehicle, at least one water-insoluble swellable mucoadhesive polymer, at least one pH-

The pH-sensitive mucoadhesive wax-film composite is a bi-layer pharmaceutical film less than 5 mm in diameter that when applied to the skin or mucosal surface adheres to the application for at least 1 hour. The wax-film composite is comprised of a pH-sensitive mucoadhesive layer and a water-insoluble wax layer.

As used herein, certain terms may have the following defined meanings.

The term *molecule of interest* is defined as any synthetic or naturally occurring substance including elements, radioactive elements, synthetic or natural small molecules, peptides, proteins, nucleic acids, or any combinations thereof.

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Amlexanox: The chemical name of Amlexanox is 2-amino-7-(1-methylethyl)-5-oxo-5H-[1]benzopyrano-[2,3-b]-pyridine-3-carboxylic acid. Amlexanox is the active ingredient in the commercial product, Aphthasol, which accelerates the healing of aphthous ulcers in the mouth by an unknown mechanism (Binnie *et al.* 1997; Khandwala *et al.* 1997a; Khandwala *et al.* 1997b). The Aphthasol product is a 5% paste. In cell culture experiments, Amlexanox has been shown to inhibit the formation or release of inflammatory mediators such as histamine and leukotrienes from mast cells, neutrophils, and mononuclear cells (Urisu *et al.* 1990). Further, in animals, Amlexanox has anti-allergic and anti-inflammatory properties and has been shown to inhibit both immediate and delayed hypersensitivity reactions (Saijo *et al.* 1985; Saijo *et al.* 1986). Amlexanox has very poor solubility in ethanol (0.58 mg/mL) and water (0.005 mg/mL). However, the solubility of Amlexanox increases with pH.

Triclosan: The chemical name of Triclosan is 2,4,4'-Trichloro 2'-hydroxy-diphenyl-ether 5-chloro 2-(2,4-dichlorophenoxy)-phenol. Triclosan is a broad-spectrum antibacterial that has activity against a wide range of gram-positive and gram-negative bacteria (Zuckerbraun *et al.* 1998). Triclosan has been shown to inhibit bacterial fatty acid synthesis at the enoyl-acyl carrier protein reductase (FabI) step (Heath *et al.* 1999). It has gained wide-spread use as antibacterial agent that is used in toothpastes, kitchen utensils and toys. Triclosan has also been formulated in mouthwashes (Mandel 1994) and liposomes (Jones *et al.* 1993). Triclosan has a pKa of 7.9 and is insoluble in water, except in alkaline solutions where it is readily soluble (Loftsson *et al.* 1999). However, triclosan is soluble in ethanol and polyethylene glycol 400 and many other organic solvents.

Hirudin: Hirudin is the most potent and specific known inhibitor of thrombin ( $K_i = 0.2 \text{ pM}$ ) (Markwardt 1989; Markwardt 1991a; Markwardt 1991b; Fenton *et al.* 1991). Hirudin was originally isolated from medicinal leeches (*Hirudo medicinalis*), but is now available in large quantities due to recombinant techniques (Riehl-Bellon *et al.* 1989). Unlike heparin, hirudin does not require endogenous cofactors to function and does not

have associated bleeding complications. Hirudin is a relatively small 65 amino acid peptide (Mw 7000) with remarkable stability over a broad range of pH (pH 3-10), temperature (i.e., up to 60°C), and solvent conditions (Chang 1991; Eric and Caroline 1991). The main structure of hirudin is characterized by an apolar disulfide knot with a very anionic (negatively-charged) tail. The apolar knot of hirudin masks the catalytic site of thrombin by interacting with its apolar site (i.e., via hydrophobic interaction). The anionic tail of hirudin (18 amino acids) interacts with the anion-binding exosite of thrombin. It is the combination of the apolar binding and the anion-binding that accounts for the very high affinity of hirudin and thrombin. The presence of the apolar knot and anionic tail on hirudin results in both hydrophobic and hydrophilic surface interactions. For example, in solution, hirudin forms micellar complexes (e.g., tetramers) with itself. Further, the apolar region of hirudin is known to interact with cell membranes while the anionic tail remains extracellular (Fenton 1989). It is likely that the anionic portion of hirudin is electrostatically repelled from the surface of cell membranes. Novartis Pharma obtained marketing approval in the European Union in 1998 for REVASC (desirudin; recombinant hirudin) for the prevention of deep-vein thrombosis following knee and hip replacement surgery. In clinical trials, patients receiving hirudin by subcutaneous injection twice a day for 8 days had a nearly 30% lower overall rate of deep-vein thrombosis than those who received low molecular weight heparin (Eriksson *et al.* 1997). Novartis is also performing clinical trials (Phase III) for the use hirudin in acute coronary syndrome. Several other indications for hirudin are being pursued such as stasis-induced venous thrombosis, diffuse microthrombosis, and hemodialysis (Markwardt 1989). Most of the potential applications of hirudin require blood levels between 80 to 3500 ng/mL (Markwardt 1989, Markwardt 1991). A potential problem with the use of injected forms of hirudin is that there are no effective antidotes for hirudin overdosing. If upon frequent injection, the wrong dose is given, there are no methods to turn the activity of hirudin off. For this reason alone, hirudin is a good candidate for transmucosal delivery via the buccal tissue. It is envisioned that the pH-sensitive mucoadhesive film-forming gel or wax-film composite containing hirudin, can easily be applied to the buccal tissue in the mouth and provide a means to safely control the dosing and dose scheduling of hirudin.

Benzodiazepines: Benzodiazepines are among the most frequently prescribed drugs in the Western world. All benzodiazepine agonists exert similar pharmacodynamic actions (e.g., activate central GABAergic neuroinhibition, thereby inducing anxiolysis, sedation/hypnosis, anticonvulsion and muscle relaxation. In turn, benzodiazepines are used for the treatment of seizures and insomnia, preoperative or procedural sedation, and for the treatment of nausea and anxiety. Most if not all of the benzodiazepine bases have little or no solubility in water and are very lipophilic (Hjortkjaer *et al.* 1999, Bechgaard *et al.* 1997). As a result, a great deal of research has gone into developing efficacious dosage forms of these drugs including; *i*) synthesizing water-soluble derivatives such as their hydrochloride salts, *ii*) developing injectable formulations of the benzodiazepine bases that do not result in precipitation of the drugs, and *iii*) exploring alternative routes of administration including intranasal, rectal, and buccal. Previous work has shown that intranasal doses of the benzodiazepine bases dissolved in organic solvents lead to pharmacological responses with suitable pharmacokinetic profiles. However, the repeated use of organic solvents such as ethanol, polyethylene glycol, and propylene glycol for intranasal administration is not desirable. Alternatively, benzodiazepines have been given by buccal administration with very favorable results (Schwagmeier *et al.* 1998; Scott *et al.* 1999). For example, Schwagmeier *et al.* showed that the maximum plasma concentration of midazolam was 55.9 ng/mL at 30 minutes following buccal administration of 5 mg midazolam with a mean bioavailability of 74.5%. Scott *et al.* reported that the buccal administration of 10 mg midazolam per 2 mL stopped 75% (30/40) seizures in children within an average time of 6 minutes after administration. Thus, buccal administration of benzodiazepines may be an attractive alternative to injection or intranasal administration. It is envisioned that the said pH-sensitive mucoadhesive film-forming gel or wax-film composite containing a benzodiazepine can easily be applied to the buccal tissue in the mouth and provide a means to safely and reproducibly deliver doses of the benzodiazepines for the treatment of seizures and insomnia, preoperative or procedural sedation, and for the treatment of nausea and anxiety. Benzodiazepines suitable for delivery are, but not limited to, alprazolam, camazepam, clobazam, clonazepam, desmethyldiazepam, diazepam,

flunitrazepam, flurazepam, halazepam, lorazepam, lormethazepam, midazolam, nitrazepam, oxazepam, pinazepam, prazepam, triazolam, and derivatives and combinations thereof.

The term *pharmaceutically acceptable* means that the substance is generally regarded to be safe when used in such a manner that is widely acceptable by those skilled in the art. For reference of acceptable materials, refer to THE UNITED STATES PHARMACOEPIA.

The term *pH-sensitive* is defined as a substance that is affected by changes in pH so that the substance changes conformation, charge, solubility, or combinations thereof. A *pH-sensitive polymer* is defined as polymer that is affected by changes in pH so that the polymer changes conformation, charge, solubility, or combinations thereof. A *pH-sensitive film-forming polymer* is defined as polymer that is affected by changes in pH so that the polymer changes conformation, charge, solubility, or combinations thereof resulting in the precipitation, desolvation, or settling of the polymer into a film. For the purposes of this invention, it is preferred that the pH-sensitive film-forming polymers have some solubility in aqueous-based vehicles above pH 5 but limited or no solubility below pH 5. It is more preferred that the pH-sensitive film-forming polymers have some solubility in aqueous-based vehicles above pH 6 but limited or no solubility below pH 6. It is most preferred that the pH-sensitive film-forming polymers have some solubility in aqueous-based vehicles above pH 7 but limited no solubility below pH 7. Examples of pH-sensitive film-forming polymers that meet these criteria are, but not limited to, Eudragits® and cellulose acetate phthalate polymers, or derivatives thereof.

Eudragits® are synthetic cationic and anionic polymers of methacrylic acid and methacrylic acid esters in varying ratios. The polymers are generally regarded as safe and are included in the FDA Inactive Ingredients Guide (for oral capsules and tablets). The presence of positive charge (provided by the dimethylaminoethylmethacrylates), neutral charge (provided by the methacrylic acid esters) and negative charge (provided by the methacrylic acid) residues provides for interesting and useful solubility characteristics as a function of pH. In general, the cationic Eudragits are soluble at lower pHs while the anionic Eudragits are soluble at higher pHs. Neutral Eudragits usually lack aqueous solubility at any pH. For the purposes of this invention, any Eudragit polymer or derivative



thereof is preferred. Even more preferred is any anionic Eudragit or derivative thereof. Most preferred is Eudragit L100-55, Eudragit L100, and Eudragit S100 or derivatives thereof. Depending on the type of Eudragit polymer selected and its solubility as function of pH, films having different properties can be prepared. Eudragit gels have been reported for the sustained-release of drugs via rectal administration (Goto *et al.* 1991; Kawata *et al.* 1991; Kim *et al.* 1992a; Kim *et al.* 1992b; Kim *et al.* 1992c; Umejima *et al.* 1993). Eudragit has also been used in buccal patches (Wong *et al.* 1999). Wong *et al.* described the use of Eudragit patches with the water-soluble cellulose-based polymers and swellable Carbopol polymers to deliver a model drug. However, for these studies Wong *et al.* utilized Eudragit NE40D, a neutral poly(ethylacrylate methylmethacrylate) polymer that is insoluble at all pHs (non-pH-sensitive). Eudragit has also been used to film-coat tablets for pH-dependent colon targeting after oral delivery (Khan *et al.* 1999). Cationic Eudragit E, which is soluble at low pH, has also been proposed for the pH-controlled drug release from mini-tablets for the treatment of inflammatory bowel disease (Leopold and Eikeler, 1998). No references or other prior art could be found that describe the use of mucoadhesive gels comprised of anionic polymers, that when applied to the skin or mucosal surfaces, form films due to effect of pH on the solubility of the anionic polymers.

*Solubility* refers to the extent to which a solute is dissolved in a solvent. Solubility can be described in terms such as described in REMINGTON'S PHARMACEUTICAL SCIENCES 18th edition, 1990, ranging from very soluble (less than 1 part of solvent per 1 part of solute) to insoluble (more than 10,000 parts of solvent for 1 part of solute). The term *water-insoluble* refers to a substance or solute where more than 10,000 parts of water are needed to dissolve 1 part of solute. A water-insoluble swellable substance refers to a substance or solute that is water-insoluble but absorbs water and swells to form a colloidal dispersion. A *colloidal dispersion*, as defined by REMINGTON'S "consists of at least two discrete phases", namely a solid phase and a liquid phase. The term *desolvation* refers to phenomena whereby a solvent such as water diffuses from a swelled substance leaving primarily one solid phase comprised of the substance. A *gel* is defined as a semisolid consisting of particles interpenetrated by a liquid. A *gel solution* is defined as a gel which

has a molecule of interest dissolved in solution in the liquid phase. A *gel suspension* is defined as a gel which has a molecule of interest suspended in the liquid phase.

A *wax* refers to any water-insoluble substance composed of hydrocarbons, alcohols, fatty acids, and esters that are solids at temperatures below 40°C, but liquids at temperatures of above 40°C. Suitable waxes are, but not limited to the following, DENTSPLY® Utility Wax, beeswax, emulsifying wax, microcrystalline wax, carnauba wax, paraffin wax, white wax, yellow wax, or other suitable pharmaceutical wax. It is preferred that the wax has a melting temperature between 40°C and 100°C, more preferred that the wax has a melting temperature between 40°C and 80°C, and most preferred that the wax has a melting temperature between 40°C and 65°C.

A *wax-film composite* refers to a bi-layer film comprised of a pH-sensitive mucoadhesive layer and a water-insoluble wax layer. A molecule of interest may be loaded in and released from either the said pH-sensitive mucoadhesive layer or said water-insoluble wax layer. It is preferred that the molecule of interest is present in the wax-film composite at a concentration of 0.001% to 20% by weight, more preferably at a concentration of 0.001% to 10% by weight, and most preferably at a concentration of 0.001% to 5% by weight. It is preferred that the pH-sensitive mucoadhesive layer is present at a concentration in the wax-film composite of 20% to 90% by weight, more preferably at 30% to 80% by weight, and most preferably at 40% to 70% by weight. Preferably, the water-insoluble wax layer is present at a concentration in the wax-film composite of 10% to 80% by weight, more preferably at 20% to 70% by weight, and most preferably at 30% to 60% by weight. For bonding the two layers of the bi-layer wax-film composite, it is preferred that the water-insoluble wax layer contain at least one water-soluble or water-swellaable agents such as, but not limited to, tragacanth, polyvinyl pyrrolidone, polyvinyl alcohol, cross-linked polyacrylic acid, polyethylene glycol, a cellulose-based polymer or derivative thereof, a cross-linked polyacrylic acid polymer or derivative thereof, or other suitable pharmaceutical polymers that are water-soluble or water-swellaable. Preferably, the water-soluble or water-swellaable agents are present in the wax layer at a concentration of 0.05% to 10% by weight, more preferably at 1% to 8% by weight, and most preferably at 1% to 5% by weight. It is envisioned that the wax-film

The term *mucoadhesive* refers to a substance that sticks to or adheres to the skin or mucosal surfaces by forces that are measurable and by any number of mechanisms such as, but not limited to the following: hydrogen-bonding, ionic interaction, hydrophobic interaction, van der Waals interaction, or combinations thereof.

Transmucosal Delivery: Peptides, proteins, and other larger molecules are becoming increasingly more important as scientists search for new drugs to treat serious human diseases. Most often, these new larger molecules have to be injected since bioavailability by other routes is low. However, developing injectable large molecule drugs is expensive and patient compliance is often low. Researchers in the field of drug delivery have sought for decades for methods to dose larger molecules like peptides and proteins orally. However these drugs are prone to rapid and extensive degradation due to the low pH of the stomach as well as to peptidases and proteolytic enzymes throughout the gastrointestinal tract. Alternative routes have been sought such as transmucosal delivery via the lung, nose, and buccal tissue. The foreseen advantages of transmucosal delivery are; *i*) a faster onset of activity (i.e., achieve higher drug blood levels faster), *ii*) the

avoidance of the first hepatic pass metabolism and the related increase of bioavailability, and *iii*) the avoidance of side effects related to GI absorption (Rathbone *et al.* 1996). However, the barriers for effective transmucosal delivery (i.e., buccal delivery) are similar to those for oral delivery. For example, the oral mucosal tissue is also characterized by a mucous layer that lines the epithelia. The mucous layer provides a substantial physical barrier as well as a chemical barrier due to the presence of a multitude of peptidases and proteolytic enzymes. In theory, delivery systems designed to adhere to this mucous layer (i.e., mucoadhesives) can serve to protect peptides from rapid and/or extensive degradation (Bodde *et al.* 1990; Duchene *et al.* 1988; Harris and Robinson 1990). The mechanism of mucoadhesion is complicated but is known to involve hydrogen bonding between the components of the delivery systems (i.e., polymers) and sialic acid residues of mucin (0.5-2.0% glycoprotein), the primary substance of mucous. The viscosity of the mucous layer is largely determined by the type and amount of the glycoproteins present. Usually, the mucous layer in the oral mucosa is about 500-600  $\mu\text{m}$  thick and has a pH of 5.8-7.1. The combination of the mucous layer and the epithelial layer provides a substantial barrier for significant drug absorption. This is especially true for peptide drugs that are prone to a high degree of proteolytic degradation in the mucous layer. Further, if the peptide drugs are anionic, the negatively-charged epithelial membrane below the mucous provides a repulsive electrical "fence" of sorts.

In general, the oral transmucosal bioavailability of peptides is low (<5%), but can be increased with the use of penetration enhancers designed to disrupt the mucosal and epithelial membranes (Sayani *et al.* 1993; Bhatt and Johnston, 1996; Bayley *et al.* 1995; Bhatt and Johnston, 1997). The most often used types of penetration enhancers are surfactants (i.e., sodium lauryl sulfate), bile salts (i.e., sodium deoxycholate), and fatty acids (i.e., oleic acid). Although these penetration enhancers typically increase the absorption of peptides, the safety and long term use of these agents is of concern. Further, these penetration enhancers, by design, do not function to specifically protect peptides from proteolytic degradation in the mucosal membrane. Alternatively, a plausible strategy is to use more pharmaceutically acceptable excipients to physically protect peptides from rapid and extensive proteolytic degradation, rather than to rely on chemical means to disrupt

biological membranes. For example, to enhance the transmucosal delivery of negatively-charged peptides, the peptides may be complexed with pharmaceutically acceptable excipients that have positive charges such as, but not limited to, chitosan, polylysine, or protamine. It is thought that complexation will serve two functions; *i*) decrease electrostatic repulsion of the peptide drugs with the negatively-charged epithelial membrane, and *ii*) decrease the extent of degradation of the peptide drugs by proteolytic enzymes in the mucosal membrane.

A *complex* refers to the physical interaction between two substances whereby a definite stoichiometry exists. Complexation can occur by any number of mechanisms, including, but not limited to: hydrogen-bonding, ionic interaction, hydrophobic interaction, van der Waals interaction, or combinations thereof. It is envisioned that molecules of interest formulated in said delivery systems may be complexed with various pharmaceutically acceptable agents to enhance their efficacy by any number of mechanisms described above, or those still unknown.

Chitosan: Chitosan is a biodegradable polysaccharide composed of two subunits, D-glucosamine and N-acetyl-D-glucosamine linked together by  $\beta(1,4)$  glycosidic bonds. When dissolved in solution, the amino groups present in the glucosamine subunits (pKa 6.5) have a positive charge (Henriksen *et al.* 1997; Schipper *et al.* 1996). Chitosan is known to be a mucoadhesive, and interacts ionically with the negatively charged sialic acid residues in mucin (Lehr *et al.* 1992). Chitosan has been used as a component of mucoadhesive systems (Takeuchi *et al.* 1996), as well as a controlled release and drug carrier systems (Berthold *et al.* 1993; Shiraishi *et al.* 1993). Chitosan has also been found to increase the nasal absorption of insulin and calcitonin (Illum *et al.* 1994; Aspdén *et al.* 1996), although the mechanism by which this occurs is not clear. Finally, chitosan has been reported to form complexes with many negatively-charged molecules including indomethacin (Imai *et al.* 1991), sodium hyaluronate (Takayama *et al.* 1990), and pectin and acacia (Meshali *et al.* 1993). One problem that may be encountered with the use of chitosan (100 kDa) is the poor solubility of 100 kDa chitosan at neutral pH. This lack of solubility may hinder its ability to form complexes with hirudin or our ability to study the complex. A solution to this problem may be to obtain lower molecular weight chitosans

(i.e., 10 kDa or 50 kDa). However, lower molecular weight chitosans are not commercially available. In our previous work with chitosan (MacLaughlin *et al.* 1998), we obtained lower molecular weight chitosans using a modified depolymerization technique first described by Peniston *et al.* (1975). This depolymerization process is a relatively simple deamination reaction where chitosan is chemically treated with sodium nitrite in 6% acetic acid at 25°C for 1 hour. The amount of sodium nitrite used in the reaction determines the resulting chitosan molecular weight since the reaction is stoichiometric. For example, reacting 1 g of 100 kDa chitosan (Seacure 143) with 2% and 7% sodium nitrite results in final chitosan molecular weights of 50 kDa and 10 kDa, respectively. The depolymerized chitosan polymers are purified and ready to use. The molecular weight of each depolymerized chitosan polymer is determined by gel chromatography through a Beckman Ultraspherogel SEC2000 column with a 1 mL/min flow-rate of eluent 0.1M acetic acid/sodium acetate buffer containing 0.05M sodium chloride as the mobile phase.

Mucosal Vaccines: It is thought that the most effective vaccines for important viral and bacterial pathogens will require mucosal immunity since this is the site at which these pathogens infect the body (McGhee *et al.* 1992). However, all marketed vaccines except one (oral polio vaccine) are now administered by injectable routes and are ineffective at inducing mucosal immunity. A new promising flu vaccine (Flumist™ from Aviron) is currently being tested in phase III clinical trials. Flumist™ is administered as a nasal spray. Since the nasal associated lymphoid tissue (NALT) is part of the mucosal network, this vaccine is the first of potentially many new mucosal vaccines. Other mucosal routes are actively being pursued by researchers including the vaginal route and the mucosal regions of the gut (i.e., Peyer's patches). However, the one mucosal route that has been largely overlooked by researchers is the oral mucosal region of the mouth. This may be due to the lack of suitable delivery systems to retain and protect antigen in the mouth for an extended period of time. We hypothesize that the oral mucosa may be an ideal site for vaccination for the following reasons: *i*) the oral mucosa is accessible, *ii*) administration to the oral mucosa is needleless, and *iii*) the oral mucosa is a preferred site for antigen presentation. Vaccination without the use of needles would provide a distinct advantage in terms of both cost and safety over conventional vaccines that must be given with needles. The oral

mucosa is theoretically a preferred site for antigen presentation since in the normal course of a day our bodies are exposed to hundreds of antigens during eating and other hand-to-mouth activities. In most cases, the oral mucosal tissue, which comprises a high density of Langerhans cells, and T-lymphocytes as well as the mucosal associated lymphoid tissue (MALT), is our first line of defense against both viral and bacterial pathogens. The percentage of Langerhan cells in the oral mucosa, about 4% of the total cells, is comparable to the percentage of Langerhan cells found in the skin epidermis. Langerhan cells are immature dendritic cells that are thought to be the most potent antigen presenting cell type. However, the oral mucosa has about a 40-fold increase in the number of T-lymphocytes as compared to the skin. The presence of Langerhan cells and T-lymphocytes along with MALT tissue (i.e., tonsils, salivary glands, Waldeyer's Ring, and the pharyngeal lymphoid tissue) makes the oral mucosa a theoretically potent immunization site. For example, unlike vaccination by the skin, intravenous, subcutaneous, and muscle routes, vaccination via the oral mucosa route may result in both cellular and mucosal (humoral) immune responses. Finally, vaccination by the oral mucosa route to target the MALT remains a relatively untested strategy as compared to previous efforts to target the nasal-associated lymphoid tissues (NALT) or the gut-associated lymphoid tissue (GALT). One explanation for the lack of published results on vaccination by the oral mucosa route is the lack of suitable pharmaceutical formulations to retain and stabilize antigens at the site for a suitable length of time.

Oral Mucosal Delivery and Vaccination: Many reports in the literature have described the use of polymeric delivery systems to deliver drugs locally in the mouth or transmucosally into the systemic circulation. See Leung *et al.* (1991) for a review on delivery systems for the oral mucosa. However, to our knowledge no group has reported on the use of such delivery systems to deliver antigens, specifically DNA plasmids expressing antigens, in the oral mucosa. Two recent published studies have investigated the application of antigens alone (with no delivery systems) to the oral mucosa route for immunization. Etchart *et al.* (1997) showed that the use of a DNA plasmid encoding measles virus hemoagglutinin (HA) injected into the buccal mucosa of mice induced HA-specific cytotoxic-T-lymphocyte (CTL) responses. The CTL responses to expressed HA

after DNA plasmid was applied to the buccal mucosa were much stronger than those responses after oral or intrajejunal immunization suggesting the feasibility of the oral mucosa as a route of immunization. Also, Casetta and Negretti (1998) demonstrated that antigenic mixtures of several bacteria could induce specific salivary IgA after rubbing the mixtures on the rat gingival mucosa. These publications demonstrate the potential utility of the oral mucosa as a route of immunization for both genetic vaccines (i.e., DNA plasmids expressing antigens) and subunit (protein antigen) approaches.

Genetic Vaccines: After the concept of genetic immunization was first demonstrated by Johnston's group in 1992 (Tang *et al.*, 1992), numerous studies have reported the potential prophylactic and therapeutic use of genetic vaccines for combating various infectious diseases (Taubes, 1997; Ulmer *et al.*, 1996; Robinson and Torres, 1997; Liu *et al.*, 1995). Vaccines of this composition appear to be both efficacious in the short term and able to elicit a prolonged anamnestic response, which is capable of preventing or resolving infection when challenged up to one year after vaccination (Yankauckas *et al.*, 1993). Genetic vaccines elicit a broader immune response than do subunit vaccines. They induce both cellular and humoral responses that are reminiscent of attenuated and whole-killed viral vaccines. Further, genetic vaccines can be prepared with relative ease of synthesis and production. Expression plasmids can be generated quickly once the antigen's coding sequence is known and small- and large-scale DNA purification methods are well established. Genetic vaccines also avoid some of the safety concerns of conventional vaccines in that there is no chance of disease due to co-purification of contaminating virus or reversion of the attenuated strain in the patient.

Vaginal Delivery: Many types of dosage forms have been proposed for the vaginal delivery of drugs including gels, creams, suppositories, films, foams, and tablets. In designing a dosage form intended for vaginal application, one must consider multiple parameters including, but not limited to, *i*) the physico-chemical properties of the drug (i.e., solubility, stability, etc.), *ii*) the intended duration of action or activity, *iii*) the context in which the dosage form is used, and *iv*) patient preferences. For example, clinical studies have shown that a majority of women prefer a gel to a cream formulation and both of these over the other dosage forms listed above (Hardy *et al.* 1998a). Women also prefer a dosage



form that has no odor or a pleasant mild odor and also one that has no color (Hardy *et al.* 1998b). The gel formulations will be able to be applied manually or by use of a simple applicator. Importantly for use a microbicidal agents, the applied gel formulations will be able to cover a large area of the mucosal tissue, adhere to the tissue, and potentially provide prolonged pharmacological activity of the molecule of interest.

A plethora of different mucoadhesive gel formulations have been described in the literature and several are now marketed products. Typically, the mucoadhesive component of the formulation is a biocompatible polymer, such as polyacrylic acid that is cross-linked with an acceptable agent to create an insoluble gel. The use of an insoluble gel is desirable since it remains in contact with the mucosal tissue for much longer periods of time. Cross-linked polyacrylic acid polymers, such as Noveon and Carbomer, have been shown to stay attached to the mucosal lining in the vagina for up to three to five days (March and Nakamura, 1993). Further, gels containing Noveon and/or Carbomer have been used as vaginal lubricants so it is envisioned that the described gels may be used during sexual intercourse. Noveon and Carbomer-based polymers are weak acids and contain many negatively-charged carboxyl-groups. The multiple negative charges on these polymers promote hydrogen-bonding between the polymers and the negatively charged mucin, the main glycoprotein that allows for the attachment of mucus to the epithelial lining of the vaginal wall (Park and Robinson, 1985). Noveon and Carbomer-based polymers have been shown to have maximum hydrogen-bonding in the pH range of 4.0 to 6.0. This is ideal for use in the vagina which has a normal pH value of about 4.5 (Stevens-Simon *et al.*, 1994; Garcia-Closas, *et al.*, 1999). It is envisioned that gels comprised of pH-sensitive polymers and water-insoluble mucoadhesive polymers such as Noveon and Carbomer may provide superior delivery of molecules of interest to the vagina since the lower pH of the vagina will cause the pH-sensitive polymer to form a long lasting film to retain and/or deliver the molecule of interest in a more efficacious manner.

The following Examples are described to illustrate the different aspects of the invention. These Examples should not be used in any way to limit the scope of the invention.

## **EXAMPLES**

**Materials:** Recombinant hirudin and bovine  $\alpha$ -thrombin are from Sigma Chemicals (St. Louis, MO). Chromozym-TH is from Boehringer Mannheim. Chitosan Seacure 143 (85-90% deacetylated) is from Natural Biopolymer Inc. (Raymond, WA). All Eudragit polymers were obtained from Rohm America, Inc. (Piscataway, NJ). Noveon and Carbomers were obtained from BF Goodrich (Cleveland, Ohio). Glycerin, polyethylene glycol 400, isopropyl myristate, ethanol, sodium hydroxide, and propylene glycol were all of USP/NF grade and were purchased from Spectrum Quality Products, Inc. (New Brunswick, NJ). PicoGreen dsDNA Quantitation Kit was purchased from Molecular Probes, Inc. (Eugene, OR). DENTSPLY® Utility Wax was obtained from DENTSPLY International (York, PA).

### **EXAMPLE 1**

A placebo pH-sensitive mucoadhesive film-forming gel was made as follows. Water (44.1 % w/w) was added to a 250 mL stainless steel beaker and stirring was begun at 200 rpm using a Caframo Stirrer. Noveon (0.5% w/w) and Carbomer 971 (0.8% w/w) were added very slowly to the stirring water until the solution was clear and viscous with no visible solid material in solution. Glycerin (50.4% w/w) was then added to the polymers in water. Eudragit L100 (2.0% w/w) was added and the solution and the viscous solution became slightly milky in color and less viscous. 18% sodium hydroxide (2.2% w/w) was then added and the whitish gel became viscous. The pH of the placebo gel was measured by taking 1 mL of the gel and dispersing it into 5 mL water and measuring the pH after 1 hour. The pH of the gel was  $6.3 \pm 0.02$  ( $n = 3$ ). When the placebo gel was spread onto the skin of a human volunteer's hand; it produced a clear film. Placebo gel stored under controlled conditions at 25°C/60% Relative Humidity for 1 week and 1 month had pH of  $6.2 \pm 0.07$  ( $n = 3$ ) and  $6.2 \pm 0.06$  ( $n = 3$ ), respectively. Placebo gel stored under controlled conditions at 40°C/75% Relative Humidity for 1 week and 1 month had pH of  $6.1 \pm 0.03$  ( $n = 3$ ) and  $6.1 \pm 0.09$  ( $n = 3$ ), respectively. These results demonstrated that the placebo gel was stable when stored under the conditions tested.

**EXAMPLE 2**

A 5% triclosan pH-sensitive mucoadhesive film-forming gel suspension was made as follows. Water (39.0% w/w) was added to a 250 mL stainless steel beaker and stirring was begun at 200 rpm using a Caframo Stirrer. Noveon (0.5% w/w) and Carbomer 971 (0.65% w/w) were added very slowly to the stirring water until the solution was clear and viscous with no visible solid material in solution. Glycerin (50.6 g) was then added to the polymers in water. Eudragit L100 (2.0% w/w) was added and the solution and the viscous solution became slightly milky in color and less viscous. 18% sodium hydroxide (2.4% w/w) was then added and the whitish gel became viscous. Triclosan (5.0% w/w) was then added to produce a homogenous whitish gel suspension. The pH of the 5% triclosan gel was measured by taking 1 mL of the gel and dispersing it into 5 mL water and measuring the pH after 1 hour. The pH of the gel was  $6.5 \pm 0.1$  ( $n = 3$ ). When the 5% triclosan gel was spread onto the skin of a human volunteer's hand, it produced a clear film. The 5% triclosan gel was stored under controlled conditions at 25°C/60% Relative Humidity for 1 week and 1 month had pH of  $6.5 \pm 0.1$  ( $n = 3$ ) and  $6.4 \pm 0.1$  ( $n = 3$ ), respectively. 5% triclosan gel stored under controlled conditions at 40°C/75% Relative Humidity for 1 week and 1 month had pH of  $6.3 \pm 0.02$  ( $n = 3$ ) and  $6.2 \pm 0.1$  ( $n = 3$ ), respectively. These results demonstrated that the 5% triclosan gel was stable when stored under the conditions tested.

**EXAMPLE 3**

A 1% triclosan pH-sensitive mucoadhesive film-forming gel solution was made as follows. Propylene glycol (37.6% w/w) was added to a 250 mL stainless steel beaker and stirring was begun at 200 rpm using a Caframo Stirrer. Triclosan (1.0% w/w) was added to the stirring propylene glycol to produce a clear semi-viscous solution. Eudragit L100 (0.2% w/w) was then added to the solution. Polyethylene glycol 400 (30.8% w/w) was then added followed by 3 mM phosphate buffer, pH 8 (28.9% w/w). To the clear semi-viscous solution, Noveon (1.0% w/w) and Carbomer 971 (0.5% w/w) were added very slowly. After additional mixing for 2 hours, the final 1% triclosan gel was a very slightly opaque gel solution. When the 1% triclosan gel solution was spread onto the skin of a human volunteer's hand; it produced a clear film.

#### **EXAMPLE 4**

A placebo pH-sensitive mucoadhesive film-forming gel was made as follows. Water (94.2% w/w) was added to a 250 mL stainless steel beaker and stirring was begun at 200 rpm using a Caframo Stirrer. Noveon (1.0% w/w) and Carbomer 971 (0.5% w/w) were added very slowly to the stirring water until the solution was clear and viscous with no visible solid material in solution. Eudragit L100 (2.0% w/w) was added and the solution and the viscous solution became slightly milky in color and less viscous. 18% sodium hydroxide (1.2% w/w) was then added and the whitish gel became viscous. Propylene glycol (0.5% w/w) and polyethylene glycol 400 (0.6% w/w) were added to the viscous gel. The final placebo formulation was a white viscous gel. However, when the placebo gel was spread onto the skin of a human volunteer's hand, it produced a clear film.

#### **EXAMPLE 5**

A 5% amlexanox pH-sensitive mucoadhesive film-forming gel suspension was made as follows. Water (63.5% w/w) was added to a 250 mL glass jar and stirring was begun at 1100 rpm using a Caframo Stirrer. Carbomer 971 (0.5% w/w) was added to the stirring water and mixed until fully hydrated. Next, Noveon (1.0% w/w) was added to the clear viscous solution and mixed until fully hydrated. Glycerin (1.0% w/w) was then added followed by ethanol (21.0% w/w). To the clear viscous gel, Eudragit L100 (7.0% w/w) was added to produce a whitish gel. Amlexanox (5.0% w/w) was then added and stirring was continued to result in a homogenous suspension of Amlexanox in the viscous gel. Finally, 18% sodium hydroxide (1.0% w/w) was added to adjust the pH. The final formulation was a whitish gel suspension. When the 5% Amlexanox gel suspension was applied to the buccal region of a human volunteer's mouth, a whitish thin film formed within 20 seconds.

#### **EXAMPLE 6**

A 1.5% amlexanox pH-sensitive mucoadhesive film-forming gel solution was made as follows. A 50 mM glycine/sodium hydroxide buffer, pH 9.0 ( (75% w/w), 95% ethanol

(7.4% w/w), and glycerin (1.0% w/w) were added to a 250 mL glass jar and stirring was begun at 200 rpm using a Caframo Stirrer. Noveon (1.0% w/w) was then added to the solution and mixed at 800 rpm until fully hydrated. Carbomer 971 (1.0% w/w) was then added to the viscous gel and mixed at 800 rpm until fully hydrated. To the clear viscous gel, Eudragit L100 (0.1% w/w) was added to produce a whitish gel. A volume of approximately 7 mL 2N sodium hydroxide was added slowly to raise the pH to 8.8. At this pH, the formulation was a clear and viscous gel. Amlexanox was then added slowly in 0.25 g increments causing the pH of the whitish gel to drop to between pH 7.0 and 7.5. After each addition of Amlexanox, approximately 0.75 mL to 1.0 mL of 2N sodium hydroxide was needed to raise the pH to 8.8 at which time the gel became clear. The final concentration of Amlexanox was 1.5% w/w and the total amount of 2N sodium hydroxide added was 12 mL (13% w/w). The final formulation was a clear and viscous gel. When the 1.5% Amlexanox gel solution was spread onto the skin of a human volunteer's hand, it produced a clear film.

#### **EXAMPLE 7**

Preparation of initial wax-film composites using different waxes. A gel of Noveon (2%) and Eudragit S100 (0.5%) in ethanol was casted on a plastic circular hollow ring (diameter = 2.357 inches) fixed on Mylar film and dried in a 55°C oven to form homogenous mucoadhesive films having a diameter of 2.357 inches. Several ¼ inch disks were cut using a punch. The disks were dipped into the following melted wax compositions to form a homogeneous coating on one side of the pH-sensitive mucoadhesive films: beeswax, emulsifying wax, and DENTSPLY® Utility Wax. The results showed that the dried beeswax and emulsifying wax layers peeled away from the mucoadhesive layer over 30 minutes. The DENTSPLY® Utility Wax exhibited a very slight tendency to peel away but was mostly adhered. In an effort to mold the two layers of the wax-film composites more efficiently, a water-swellable polymer, tragacanth, was dispersed into melted DENTSPLY® Utility Wax at a concentration of either 1% w/w, 7% w/w, and 10% w/w. Mucoadhesive disks were then dipped into the new melted wax compositions to form a homogeneous coating on one side of the mucoadhesive films. The

results showed that the two layers, mucoadhesive layer and wax layer with 1% tragacanth, were permanently molded together when stored at ambient conditions. Wax-film composites made with either 7% w/w or 10% w/w tragacanth in the wax layer caused the wax to be too permeable to water.

### **EXAMPLE 8**

Six different pH-sensitive mucoadhesive wax-film composites were made using six different Noveon and Eudragit S100 compositions in ethanol as shown in Table 1 below. A specific volume of each gel was added to a plastic circular hollow ring (diameter = 2.357 inches) fixed on Mylar film to result in 450-480 mg total polymer in the ring.

**Table 1: Preparation of Wax-Film Composites Having Different Ratios of Noveon/Eudragit S100 in the pH-Sensitive Mucoadhesive Layer**

Gel #	Noveon (%) in gel	Eudragit (%) in gel	Volume of Gel Added to Ring (mL)	Noveon (mg) in Dried Film	Eudragit (mg) in Dried Film	Ratio of Noveon/Eudragit in pH-sensitive mucoadhesive Wax-film composite
1	1	2	15	150	300	1:2
2	2	2	12	240	240	2:2
3	2	1	15	300	150	2:1
4	1	0.5	30	300	150	2:1
5	2	0.67	12	240	113.9	3:1
6	2	0.5	18.5	370	92.5	4:1

The ethanol from each Gel 1-6 was evaporated in a 55°C oven to form homogenous films (Films 1-6) of having a diameter of 2.357 inches. The dry Films 1-6 still on the Mylar film were dipped in melted DENTSPLY® Utility Wax (containing 1% w/w tragacanth polymer) at 55°C resulting in a homogenous coating of a wax layer on one side of the pH-sensitive mucoadhesive wax-film composites 1-6. Three-¼ inch disks each of the wax-film composites 1-6 for were made using a ¼ inch punch. Exactly 6 µL of a 40 mM KH<sub>2</sub>PO<sub>4</sub>/NaOH Buffer, pH 6 was added to the mucoadhesive side of the ¼ inch wax-film composites 1-6 (n = 3) and the disks were fixed to glass microscope cover-slips (to

### EXAMPLE 9

**Table 2: Preparation of Wax-Film Composites Having Different Total Polymer Contents (Noveon/Eudragit S100) in the pH-Sensitive Mucoadhesive Layer**

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The ethanol from each Samples 1-5 was evaporated in a 55°C oven to form homogenous films (Films 1-5) having a diameter of 2.375 inches. The dry Films 1-5 still on the Mylar film were dipped in melted DENTSPLY® Utility Wax (containing 1% w/w tragacanth polymer) at 55°C resulting in a homogenous coating of the pH-sensitive mucoadhesive wax-film composites 1-5. Three-¼ inch disks each of the wax-film composites 1-5 for were made using a ¼ inch paper punch. Exactly 6 µL of a 40 mM KH<sub>2</sub>PO<sub>4</sub>/NaOH Buffer, pH 6 was added to the mucoadhesive side of the ¼ inch wax-film composites 1-6 (n = 3) and the disks were fixed to glass microscope cover-slips (to simulate the negatively-charged mucosal surface). The disks were allowed to dry for 4 hours at room temperature and then each cover-slip containing adhered disk was placed in a bath of 40 mM KH<sub>2</sub>PO<sub>4</sub>/NaOH Buffer, pH 6 at 37°C (stirred at 100 rpms). The time at which each disk dislodged from the cover-slip was recorded and the results of the study are plotted in Figure 3. The data demonstrated that when the amount of total polymer per ¼ inch disk was increased, there was no effect on the adhesive time on the glass. This result was most likely due to the fact that both the glass and the wax layer are impermeable to the buffer. It is expected that if the wax-film composites as described in this Example were placed on a permeable substrate such as a membrane or the buccal tissue, differences in the adhesion time would be observed.

#### **EXAMPLE 10**

Pre-loading and post-loading of plasmid DNA into pH-sensitive mucoadhesive wax-film composites was carried out. Plasmid DNA (CMV-luciferase plasmid) was loaded into the wax-film composites during the manufacturing process or post-loaded onto pre-made composites. An ethanol-based gel containing 3% Noveon and 1% Eudragit S-100 was prepared. For pre-loaded composites, 12.6 g of the gel was diluted with 6.3 g ethanol and 400 µL of plasmid DNA (920 µg) was added. For post-loaded composites, 12.6 g of the gel was diluted with 6.3 g ethanol. Each gel was added to a plastic circular hollow ring (diameter = 2.375 inches) fixed on Mylar film. The ethanol from each gel was evaporated in a 55°C oven to form homogenous films (Pre-loaded and Post-Loaded) having a diameter of 2.375 inches. The dry Pre-loaded and Post-Loaded Films, still on the Mylar film, were



dipped in melted DENTSPLY® Utility Wax (containing 1% w/w tragacanth polymer) at 55°C resulting in a homogenous coating of the pH-sensitive mucoadhesive wax-film composites. Three ¼ inch disks each of the wax-film composites were made using a ¼ inch paper punch. Pre-loaded wax-film composites contained 10 µg plasmid DNA per ¼ inch disks. To Post-loaded wax-film composites, exactly 4.3 µL of plasmid DNA (10 µg) was added to the mucoadhesive layer and allowed to dry for 4 hours.

Adhesion time: To dry wax-film composites containing plasmid DNA (pre-loaded and post-loaded) exactly 6 µL of a 40 mM KH<sub>2</sub>PO<sub>4</sub>/NaOH Buffer, pH 6 was added to the mucoadhesive side of the ¼ inch wax-film composites (n = 3 for both Pre-loaded and Post-loaded) and the disks were fixed to the inside of a 300 mL glass beaker (to simulate the negatively-charged mucosal surface). The disks were allowed to dry for 4 hours at room temperature and then submerged in 40 mM KH<sub>2</sub>PO<sub>4</sub>/NaOH Buffer, pH 6 at 37°C (stirred at 100 rpms). The time at which each disk dislodged from the glass was recorded and the results of the study are plotted in Figure 4. The data demonstrated that pre-loading or post-loading plasmid DNA had no effect on the adhesion times of the wax-film composites to glass.

Quantitation of Plasmid DNA: Plasmid DNA in solution was quantified using the PicoGreen dsDNA Quantitation Kit. In brief, Phage Lamda DNA was diluted with TE (Tris-EDTA) buffer (pH=7.5) to obtain a DNA concentration of 0, 2, 20, 100, and 1000 ng/mL. These solutions were further diluted twice with PicoGreen containing TE buffer. One mL of the solution was measured by the fluorescence spectrophotometer (Hitachi Instrument, Inc., Fairfield, OH) to obtain a standard plasmid DNA calibration curve as shown in Figure 5. This curve was used for calibration of the concentration of unformulated plasmid DNA and plasmid DNA released from wax-film composites.

In-Vitro Release of Plasmid DNA from Wax-Film Composites: Five ¼ inch wax-film composites containing of plasmid DNA (5 µg) (pre-loaded and post-loaded) were submerged separately into 1 mL 10 mM PBS buffer, pH 7.4 at 37°C. At various times, exactly 100 µL solution was aliquoted for DNA quantitation using the PicoGreen DNA Quantitation Kit. Exactly 100 µL fresh PBS was added to replace the removed volume at each time point. As shown in Figure 6, the release rate of plasmid DNA pre-loaded into



Hirudin was both pre-loaded and post-loaded into wax-film composites in a similar manner as described in Example 10. The release rate of hirudin (0.5 mg) from ¼ inch wax-film wax-film composites was measured using both a HPLC assay and a chromogenic substrate assay.

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when the mole ratio of hirudin to thrombin is 1. A calibration curve for released hirudin (see Figure 8B) in solution was made by plotting the difference between the thrombin base value absorbance (thrombin in the presence of substrate only) and the hirudin sample values versus the concentration of hirudin standards from Figure 8A.

HPLC assay for hirudin: Three standard curves for hirudin were generated using a C18 Nucleosil® 5 µm analytical column (4.6 x 250 mm). The mobile phase consisted of water:acetonitrile:trifluoroacetic acid (59.95:40:0.05 w/w/w). The flow rate was 0.6 mL/min. Each 50 µL sample of each standard curve was injected three times and the average AUC was plotted as shown in Figure 9.

The data suggested that wax-film composites containing either pre-loaded or post-loaded hirudin can be placed in the mouth of a warm-blooded animal to result in the absorption of hirudin into the systemic circulation in an amount suitable to elicit a pharmacological response (i.e., effect blood-clotting time).

### **EXAMPLE 12**

Hirudin/chitosan complexes were prepared by adding chitosan (from 13 µg to 125 µg) to hirudin (0.5 mg) in water so that the total volume was 1 mL. The particle size of the complexes was measured using a Coulter N4 Plus Sub-Micron Particle Sizer using 90° laser light scattering for 120 seconds. The results indicate that the complex size was effected by the weight ratio of hirudin and chitosan in solution with a hirudin/chitosan ratio of 4:1 w/w resulting in the smallest complex size of approximately 100 nm (Figure 10). Hirudin/chitosan complexes were also prepared by adding chitosan (from 2.5 µg to 25 µg) to hirudin (0.1 mg) in water so that the total volume was 1 mL. The zeta potential of the complexes was measured using a Malvern Zetasizer 2000. The zeta potential results in millivolt (mV) indicated that as the weight ratio of hirudin to chitosan increased, the overall charge of the complex decreased (Figure 11). This result was expected since hirudin is negatively-charged and chitosan is positively-charged. The particle size stability of selected complexes stored at 4°C size was also determined. Four complexes having hirudin/chitosan weight ratios of 0.1, 0.15, 0.2, and 0.25 w/w were particle sized on Day 1, 2, 6, and 7 and the results are shown in Figure 12. For the Day 7 sample analysis, exactly

All of the cited references are incorporated herein by reference in their entirety.

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